

IEA HPP Annex 34 - Nationaler Beitrag Task C

“Thermische Zersetzung und Korrosion in NH₃/H₂O Absorptionswärmepumpen”

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Motivation / Einordnung in Annex 34

Nationaler Beitrag ¹⁾

- Grundlagen und Literaturrecherche
 - a) Korrosion
 - b) Zersetzung
- Versuchsanlagen
 - a) Thermosiphonteststand
 - b) Autoklaven
 - c) Probennahme
- Experimentelle Ergebnisse
- Zusammenfassung

Bsp. für Beiträge aus anderen Ländern

- ¹⁾ Moser H., Zotter G., Kottenko O., Rieberer R. (2011): “The Formation of Non-Condensable Gases in Ammonia/Water Absorption Heat Pumps Made of Stainless Steel – Literature Review and Experimental Investigation”, Int. Conf. of Ammonia Refrigeration Technology, Ohrid, Macedonia.

Motivation / Einordnung in Annex 34



Task C: Apparatus technology

Der Task C befasste sich mit der
Komponentenentwicklung für thermisch
angetriebene Wärmepumpen mit Fokus auf
z.B.: Alterung, thermische Stabilität,
Korrosion, Inertgasbildung, Lärm usw.

Task A: Market overview/state of the art	
WP 1 – state of the art/ country reports WP 2 – Outlook WP 3 – politics/ labeling	
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THE FORMATION OF NON-CONDENSABLE GASES IN AMMONIA / WATER ABSORPTION HEAT PUMPS MADE OF STAINLESS STEEL

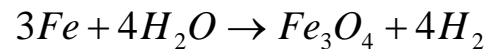
Fundamentals and Literature Review

- Corrosion
- Dissociation – Thermal Decomposition of NH₃

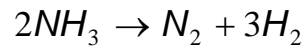
Fundamentals

High performance NH₃/H₂O absorption heat pump processes are limited by the generator temperature because of two chemical processes:

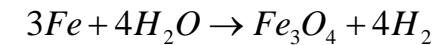
1. Corrosion of steel



2. Thermal dissociation of ammonia



Corrosion Rate



Conversion factor: ca. 20,8 g_{Fe} / g_{H₂}
 ca. 1,87 g_{Fe} / l_{H₂} (@ 1 bar, 20°C)

Example:

Surface Fe m ²	Corrosion Fe mm	H ₂ Production (@ 1bar, 20°C) l
1	0.001	4
1	0.01	42
1	0.1	422

Even small corrosion rates lead to considerable H₂-production rates!

Literature Review - Corrosion of Carbon Steel

Carbon steel and corrosion inhibitors are commonly used for NH₃/H₂O AHP

Corrosion inhibitors are sodium chromate and/or sodium dichromate.

These substances will probably be phased out in near future because of their toxic and carcinogenic nature.

Alternative inhibitors are currently under development e.g.:

- Silicon compounds
- Strong bases
- Pretreatment of the surface

Literature Review - Corrosion of Stainless Steel

Partly inconstant predictions found in literature for aqueous ammonia solution and Stainless Steel (SS)

Chemical stability brochures:

- SS is often treated as “absolutely resistant”

Dechema Corrosion Hand Book (2007)

- minor attacked at room temperature (corrosion rate < 0.05 mm/year)
- @ boiling point and/or 100°C (corrosion rate ca. 0.5 mm/year)
- > 100°C significant uniform surface corrosion

ORNL (1995):

- Stainless steel 304 < 0.002 mm/year (for 5% NH₃ @ 180°C)

No research on inhibitors with stainless steel has been found in Lit.

Literature Review - Dissociation

Contradictory information on dissociation in AHP have been found in literature e.g.:

McKelvy & Isaacs (1920):

- Hydrogen comes from corrosion process
- Nitrogen comes from air when oxygen is removed due to corrosion
- Temperatures in common AHP are too low for dissociation

Guerra (2002):

- Dissociation take place when in certain situations the temperature exceeds 300 – 350°C

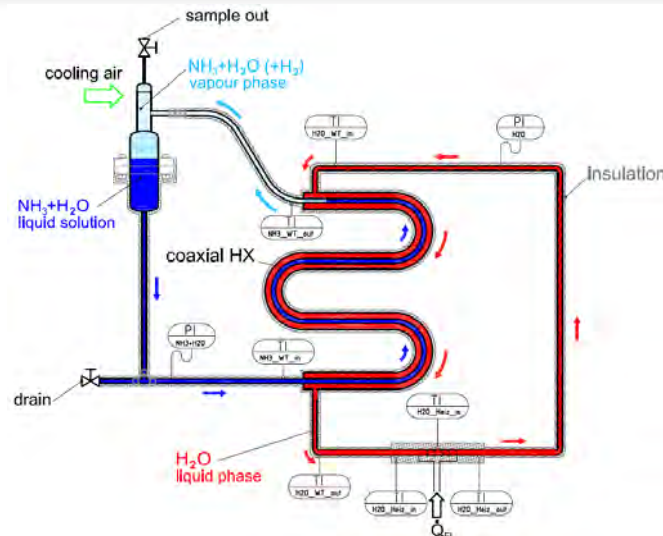
Brosby-Olsen (1996):

- Catalytic dissociation of NH_3 with Nickel as catalyst at temperatures as low as 110-120°C.

Experimental Setup

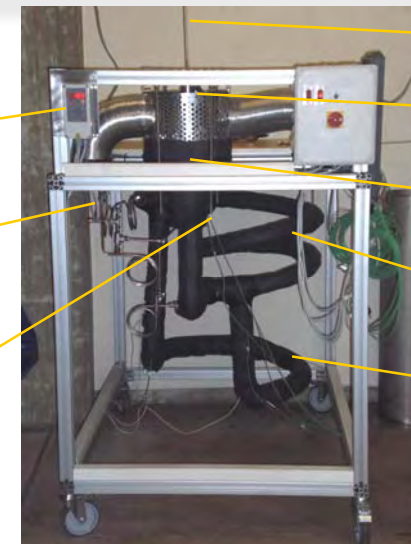
- Thermosiphon Test Rig
- Autoclaves
- Sampling and Measurement Method

Thermosiphon Test Rig:



Thermosiphon Test Rig:

Temperature control unit
 Pressure measurements
 Temperature measurements



Sampling tube
 Condenser & Absorber (not insulated)
 Space for probes of different alloys
 Generator & Bubble Pump
 Electrically heated water circuit

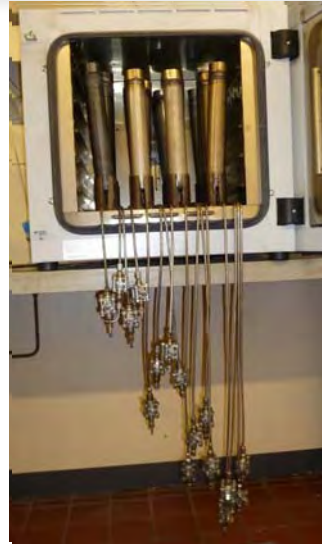
Autoclaves for Corrosion Tests

16 Test Probes have been constructed

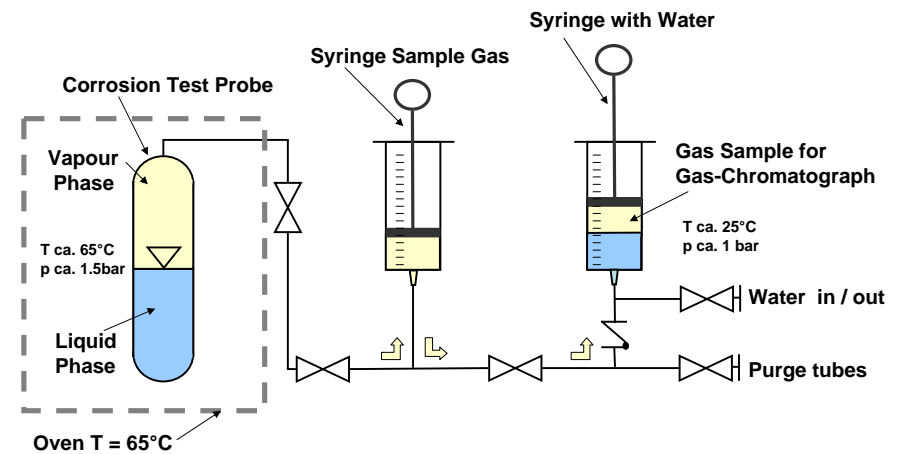
- 4 pcs. of St 37
- 4 pcs. of 1.4307
- 4 pcs. of 1.4404
- 4 pcs. of 1.4571

For stainless steel probes welding has been performed:

- without any precaution
- with forming gas only
- with acid cleaning only
- with forming gas and acid cleaning



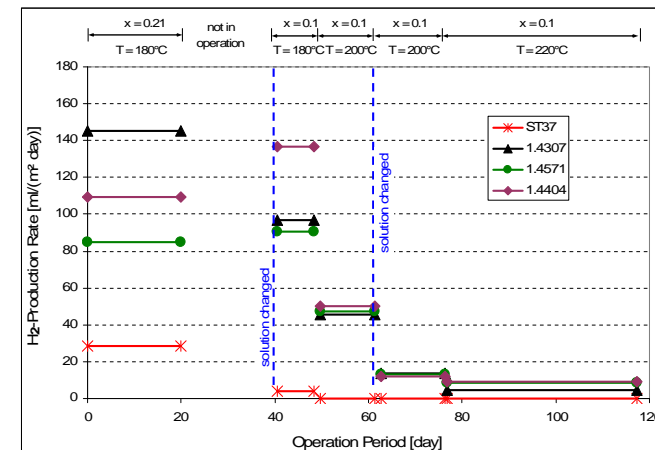
Gas Sampling and Measurement



Experimental Results

- Autoclaves
- Thermosiphon Test Rig

Results with Autoclaves

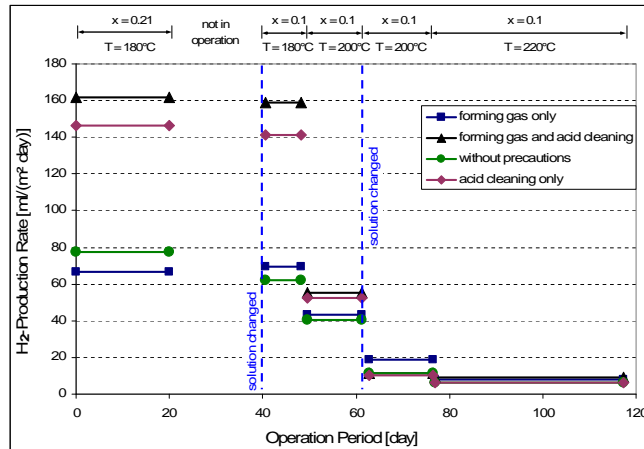


Average values for 4 probes

$X \dots [kg_{NH_3} / kg_{Solution}]$

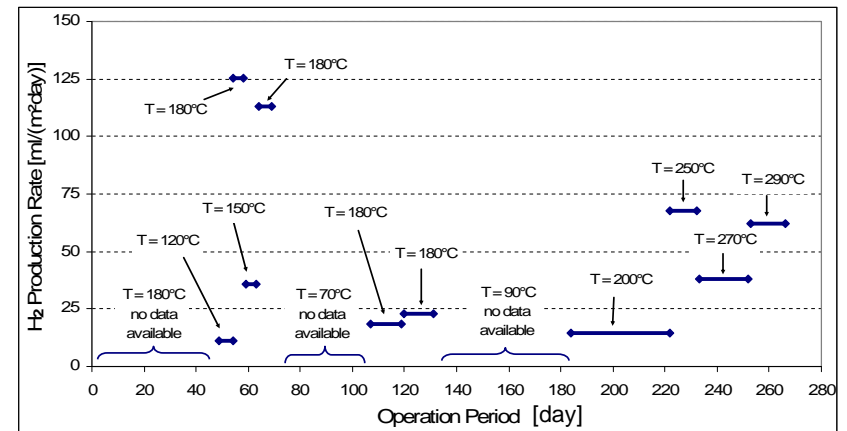
$m^2 \dots$ inner surface of probes

Results with Autoclaves



Average values of the 3 SS probes with different precaution

Results with Thermosiphon Test Rig



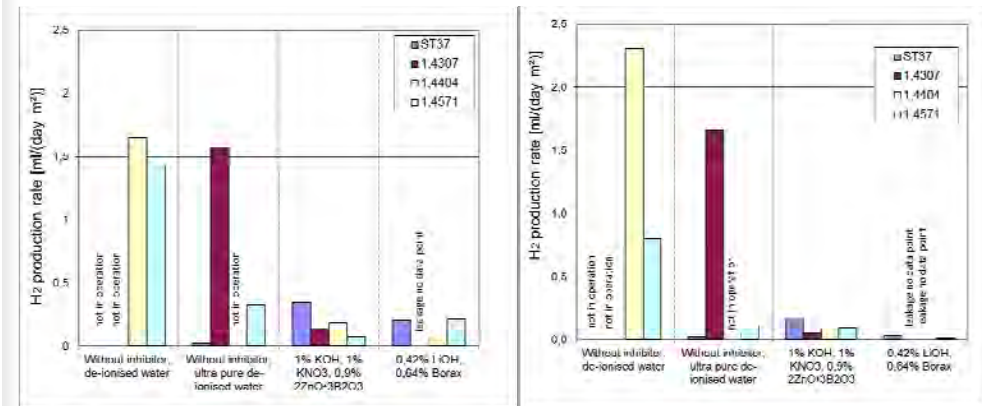
Tests with autoclaves using inhibitors

Final test with autoclaves with and without inhibitors were carried out in order to:

- observe possible thermal decomposition of ammonia ✓ **no decomposition detected!** at a temperature of 300°C
- investigate possible influence of the water quality on the corrosion and
- influence of inhibitors on the inert gas formation

Tests with autoclaves using inhibitors

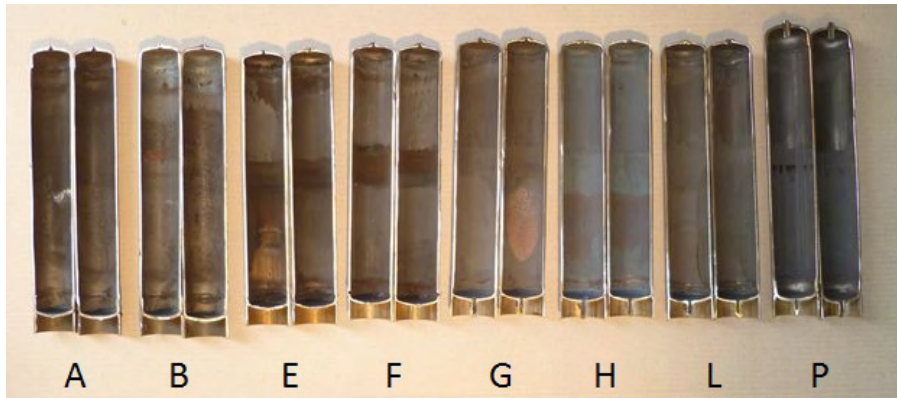
(x = 10%; t = 220°C)



22.12.2010 – 27.01.2011

10.02.2011 – 12.04.2011

Probes after Tests



All probes show thin corrosion layer – no significant difference between probes
According to calculation (based on measured amount of H₂): ca. 0.001 mm

Summary NH₃/H₂O Dissociation & Corrosion

- No thermal dissociation of ammonia has been detected up to temperatures of < 300°C
- A large initial hydrogen production rate which decreases with time has been observed
- Mild carbon steel (ST37) has shown significantly lower hydrogen production rates compared to stainless steel
- Further work is necessary in order to investigate possible precautions
- Inhibitors reduces production of inert gases slightly
- A follow up project should be initiated

Danksagung

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Danke für Ihre Aufmerksamkeit